

47 **Microscopic and mineralogical characteristics behind the engineering**
48 **properties of a compacted andesitic volcanic soil**

49

50 **Highlights**

- 51 • Drying procedure influences engineering behaviour of volcanic soils
- 52 • Key to the understanding of volcanic soil's behaviour lies in its mineralogy
- 53 • Halloysite dehydration at mineral scale explains drying's impact on behaviour

54

55 **Abstract**

56 Most of the volcanic soils found around the world are concentrated in regions of high demographic
57 and economic growth. Research shows that the drying process to which volcanic soil samples are
58 subjected to as part of the preparation process for laboratory testing can affect soil engineering
59 behaviour, inadvertently leading to undesirable results. Previous studies suggest that the key to the
60 understanding of the behaviour of volcanic soils lies in its mineralogy. Thus, this paper proposes an
61 experimental investigation on an andesitic volcanic soil from Popayan, Colombia, to investigate the
62 extent to which changes in drying affect the engineering behaviour of the soil. Mineralogical tests
63 indicated the presence of halloysite in the soil. From the geotechnical, hydromechanical and structural
64 tests it was possible to recognise that the drying procedure influenced: compaction behaviour;
65 consistency limits; particle size distribution; retention behaviour; compressibility and pore size
66 distribution ($> 1 \mu\text{m}$), while not having great effect on: specific surface; pore size distribution ($< 1 \mu\text{m}$).
67 Combined analyses of the laboratory tests performed suggested that the dehydration of halloysite at
68 mineral scale could then explain the distinct change in engineering behaviour observed when this soil
69 is subjected to drying.

70 **Key words:** volcanic soil, halloysite, hydromechanical, mineralogy, particle size distribution, pore size
71 distribution.

72

73 **1. Introduction**

74 Volcanic soils cover 1% of the Earth's surface yet they support 10% of the world's population, including
75 some of the highest human population densities (Neill, 2009; Soil Survey Staff, 1998). Andesitic

76 volcanism is at the origin of volcanic soils that rim the Pacific. They cover ample areas of Chile, Peru,
77 Ecuador, Colombia, Central America, the United States, Kamchatka, Japan, the Philippines, Indonesia,
78 New Zealand, and the independent island states of the southwest Pacific. Approximately 60% of them
79 are in regions of tropical climate (Shoji et al., 1993).

80 Early volcanic soils are composed mainly of volcanic glass; in such soils, the very rapid cooling inhibits
81 the development of a crystalline structure in the fine molten particles. This rapid cooling forms
82 distinctive clays, such as allophanes, having a “weak” degree of crystallinity (referred to as “short-
83 range order clays” or SROC). Afterward, on weathering, the amount of SROCs decreases, and the
84 amount of clays like halloysite and gibbsite increase. Then, more advanced stages of weathering
85 transform halloysites into kaolinite (Wesley, 2014).

86 Andisols are a special category of volcanic soils that have particular engineering properties, mainly
87 when they undergo weathering under tropical conditions (Neill, 2009). The main challenges regarding
88 these soils are related to the stability of natural slopes (Betancur et al., 2013; Terlien, 1997), erosions
89 (Bommer et al., 2002) and difficulties in the compaction process (Almazán, 2013; Hernandez et al.,
90 2018; Wesley, 2009). Several studies demonstrated that the geotechnical properties of volcanic soils
91 are different from region to region of origin (Betancur et al., 2013; Terlien, 1997).

92 Regarding their engineering properties, Andisols are characterized by a high macro-porosity, high
93 natural water content, and very low bulk density ($\leq 0.90 \text{ Mg m}^{-3}$). Moreover, these soils suffer marked
94 texture changes from a smeary gel at field water content to an apparent sandy soil with no cohesion
95 when air dried. Additionally, air drying also affects: soil classification (particle size distribution);
96 Atterberg limits and compaction properties (Wesley, 2014; 2009; 1973 and Hernandez et al., 2018).
97 This in turn can lead to changes in hydromechanical (Colombo et al. 2014; De Vita et al. 2012; Dorel
98 et al. 2000; Emmanuel et al. 2019; Joussein et al. 2006), structural (Bartoli et al. 2007; Emmanuel et
99 al. 2019; Hernandez et al. 2018) and cyclic (Matsumura et al. 2015) responses. Hernandez et al. (2018),
100 Colombo et al. (2014) and De Vita et al. (2012) suggested that the key to the understanding of the
101 behaviour of the volcanic soil lies in its mineralogy.

102 Kitagawa (1971) and Dörner et al. (2009) have attributed the irreversible drying and hysteresis
103 between dehydration and rehydration of volcanic ash soils observed by other researchers to the
104 allophane. Poulénard et al. (2003) observed that Andisols will become especially fragile if handled and
105 that the irreversible loss of their water holding capacity will likely induce phenomena that are harmful
106 to soil quality, according to them the largest part of the water holding capacity of these materials is
107 therefore associated with the water stored in those fine and very fine pores. Similarly, Wesley (1973)

108 reached the conclusion that the nature of the individual particles (of the halloysite) and the 'gel' (of
109 the allophane) were the main factors that contributed to the unusual engineering properties of the
110 volcanic soils investigated by the author. However, the extent to which the mineralogy plays a role on
111 the behaviour of volcanic soils still needs further investigation.

112 Thus, this paper explores this link between the mineralogy and the geotechnical change in behaviour
113 and response of an andesitic volcanic soil when samples are subjected to different drying preparation
114 procedures. Results of an experimental investigation carried out on a volcanic soil from Popayan,
115 Colombia, are presented. Firstly, mineralogical tests, were performed to assess the mineralogical
116 composition of soil samples subjected to different drying procedures when appropriate. Then, to
117 investigate the extent to which changes in drying affect the engineering behaviour of the soil,
118 geotechnical, hydromechanical and structural characterization tests were carried out on compacted
119 (following different drying preparation procedures) and intact soil samples. Only the combined
120 analyses of these findings would allow an insight into aspects that have already been discussed in
121 other publications, such as the significant variation in field and laboratory behaviour.

122

123 2. Material and methods

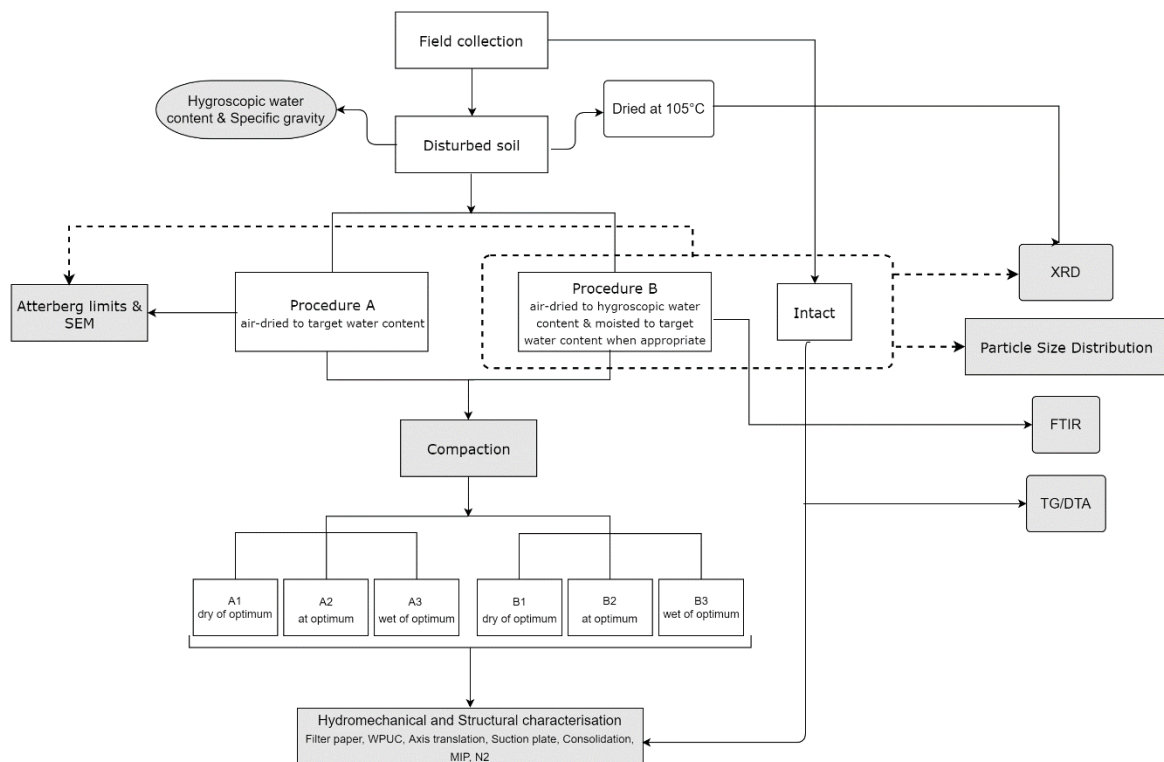
124 2.1. *Material*

125 The material used for this study was a volcanic ash from Popayán, Colombia, located 80km from
126 Puracé volcano, where the climate is classified as temperate and humid (Hurtado, 2014). Colombia's
127 volcanic ash soils from a few meters depth are typically yellowish coffee in colour, composed of highly
128 plasticity clays, originated from volcanic ash sediments deposited over 4,000 years ago (Lizcano et al.,
129 2006). For this study, samples were collected (disturbed and intact block) from a depth of 2m in
130 November, which is historically the month with the highest average precipitations.

131 2.2. *Experimental procedure and specimen preparation*

132 The experimental procedure and specimen preparation adopted are illustrated in **Fig. 1**. Three
133 different specimens were prepared: the intact specimen, moulded from the intact block. Procedure A
134 involved air drying the disturbed soil up to the target water content required, breaking soil lumps with
135 the help of a pestle and mortar, sieving the material in sieve #10 (2mm) and sealing it into a plastic
136 bag for 24hrs for homogenisation. Procedure B involved air drying the disturbed soil to the hygroscopic
137 water content, breaking soil lumps with the help of a pestle and mortar, sieving the material in sieve
138 #10 (2mm), moistening the material to the target water content when appropriate and finally sealing

139 it into a plastic bag for 24hrs for homogenisation. Procedure A is the procedure usually followed in the
 140 field, while procedure B is conventionally required by international standard for laboratory testing.

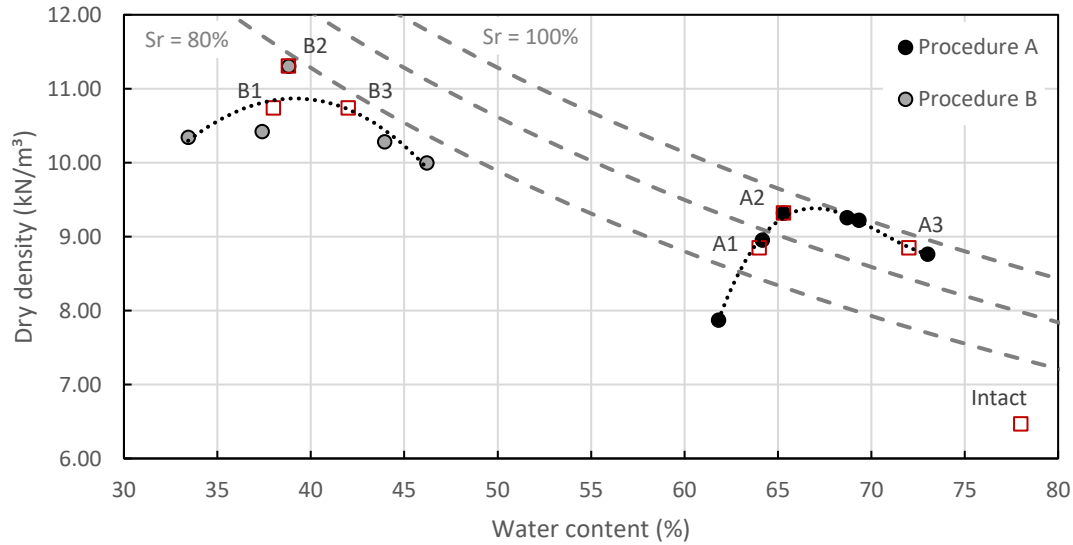


141

142 **Fig. 1.** Experimental procedure and specimen preparation.

143 The hygroscopic water content was determined according to the procedure suggested by Shah and
 144 Singh (2005) and the value obtained was 7.70% at the relative humidity and temperature of the
 145 laboratory, approximately 37% at 24 °C respectively. The specific gravity was determined following
 146 ASTM D854-14 (2014) and the value obtained was 2.59.

147 As noted by Wesley (2009) and Hernandez et al. (2018) soils of volcanic origin are sensitive to the
 148 drying process presenting a large and rapid variation in optimum water content. To analyse this
 149 variability, samples were compacted with energy corresponding to the Standard Proctor (NBR 7182,
 150 2016; ASTM D698, 2012), using specimen preparation methods A and B, as previously described (**Fig.**
 151 **1**). From these procedures, two compaction curves were obtained, as shown in **Fig. 2**.



152

153 **Fig. 2.** Standard Proctor compaction curves of procedure A and B.

154 The points of interest were defined from the two compaction curves. Three points were used for each
 155 compaction curve: on the dry of optimum (95% of the optimum dry density, A1 and B1), at optimum
 156 (A2 and B2), and on the wet of optimum (95 % of the optimum dry density, A3 and B3). **Table 1** shows
 157 the characteristics of each specimen. The porosity, degree of saturation and void ratio of the samples
 158 air dried to the hygroscopic water content (Procedure B) are lower than the samples prepared
 159 according to procedure A (without substantial drying), which demonstrates the influence of drying on
 160 the change in soil structure. **Fig. 2** also shows the dry density and water content of the intact specimen
 161 (6.47 kN/m³ and 78.0%).

162 **Table 1** Sample characteristics.

Sample	Water content, w (%)	Dry unit weight, γ_d (kN/m ³)	Initial void ratio, e_i	Degree of saturation, S_r (%)	Porosity, n	Suction, s (kPa)
Intact	78.0	6.47	2.95	68.5	0.74	298
A1	64.0	8.85	1.93	85.9	0.66	150
A2	65.3	9.32	1.78	95.1	0.64	210
A3	72.0	8.85	1.91	97.1	0.66	170
B1	38.0	10.74	1.42	69.3	0.59	29
B2	38.8	11.31	1.30	77.8	0.56	11
B3	42.0	10.74	1.42	76.6	0.59	17

163 **2.2.1. Mineralogical characterisation**

164 As a first qualitative assessment of the morphology of the material tested, Scanning Electron
 165 Microscopy (SEM) images were carried out. Specimens of the intact and compacted samples were air

166 dried to the hygroscopic water content, soil lumps were broken while the material was sieved
167 (0.075mm) manually and they were coated with gold. SEM images were performed with JSM 700 1F
168 equipment.

169 For the mineralogical characterisation, X-Ray Diffraction (XRD), Thermogravimetric Analyses (TG/DTA)
170 and Fourier Transform Infrared Spectroscopy (FTIR) tests were carried out.

171 Three specimens were prepared for XRD tests. The first one was prepared by oven drying the disturbed
172 soil at 105 °C, the second specimen was air-dried to the hygroscopic water content and the last
173 specimen was obtained from the intact block and kept moist at water content equal to 78%. Then
174 these 2 dried specimens were crushed and sieved (0.075mm), while the moist specimen was only
175 crushed to maintain the conditions found in the field. XRD analyses were carried out in the RIGAKU -
176 ULTIMA IV diffractometer, using copper tube and nickel filter. In the test, a wave with a length of 1.54
177 Å and a scanning speed of 0.05 °/min were used. From the diffractograms, minerals were identified,
178 with support from the MDI JADE 9.0 software.

179 Thermogravimetric analyses (TG/DTA) were used to study the kinetics of any phase transformation
180 processes and chemical reaction mechanisms up to 900°C with a heating rate of 10 °C/min. The
181 equipment used was the DTG-60H. TG/DTA is a technique that measures the variation in mass when
182 a material is subjected to the heating process in a controlled (inert) atmosphere. Therefore, this
183 technique does not allow identifying the mineralogical change due to the drying preparation process.
184 Hence, only specimens obtained from the intact block underwent TG/DTA testing.

185 Complementarily to the XRD and TG/DTA analyses, FTIR tests were performed using the VERTEX 70
186 equipment. A specimen obtained from the disturbed sample was air dried to the hygroscopic water
187 content crushed and sieved (0.075mm). Then, 1g of soil was mixed with 5g of Potassium bromide (KBr)
188 to prepare the tablets that were subjected to analyses. Finally, a software was used to identify the
189 absorbance spectra for each wave number, which then identifies the different existing minerals by the
190 length of the wave.

191 *2.2.2. Geotechnical characterisation*

192 Geotechnical characterisation tests included: particle size distribution, with and without dispersant
193 and Atterberg limits. Sieving and sedimentation were employed to the determination of the particle
194 size distribution. These tests were carried out on intact specimens and specimens prepared according
195 to procedure B (air-dried to hygroscopic water content) following ASTM D6913/D6913M-17 (2017) for
196 sieving and ASTM D7928 (2017) for sedimentation, with and without dispersant. The Atterberg limits
197 were determined following ASTM D4318-17 (2017) for the intact specimen and specimens that

198 underwent preparation procedure A (air-dried to 65.3% and then wetted) and procedure B (air-dried
199 to hygroscopic water content and then wetted).

200 *2.2.3. Hydromechanical characterisation*

201 The hydraulic behaviour was assessed via the Soil Water Retention Curve (SWRC). The SWRC was
202 measured for the entire suction range following drying and wetting paths. To determine the SWRC on
203 the drying path, three techniques were used: suction plate (1-15kPa), axis translation (20-200kPa) and
204 dewpoint psychrometer, WP4C (40kPa-300MPa). Before each test, specimens moulded from intact
205 and compacted (A1 to A3, B1 to B3) samples were saturated. The dimensions of specimens were:
206 2.5x1.0 cm, 6.5x2.0 cm and 5.0x0.5 cm (diameter x height) for suction plate, axis translation and WP4C,
207 respectively. For the wetting path, only WP4C technique was used.

208 Compressibility was evaluated through consolidation tests, carried out according to ASTM
209 D2435/D2435M (2020). Specimens of 5.0 cm in diameter and 2.0 cm in height moulded from intact
210 and compacted (A1 to A3, B1 to B3) samples were first saturated and then subjected to successive
211 increments of vertical stress, starting from 10kPa. Stresses were doubled at each new step up to
212 1,200kPa. Unloading steps, back to 10kPa, took place in similar fashion.

213 *2.2.4. Structural characterisation*

214 To assess the structural behaviour, porosimetry tests were carried out by adsorption and desorption
215 of nitrogen and Mercury Intrusion Porosimetry (MIP).

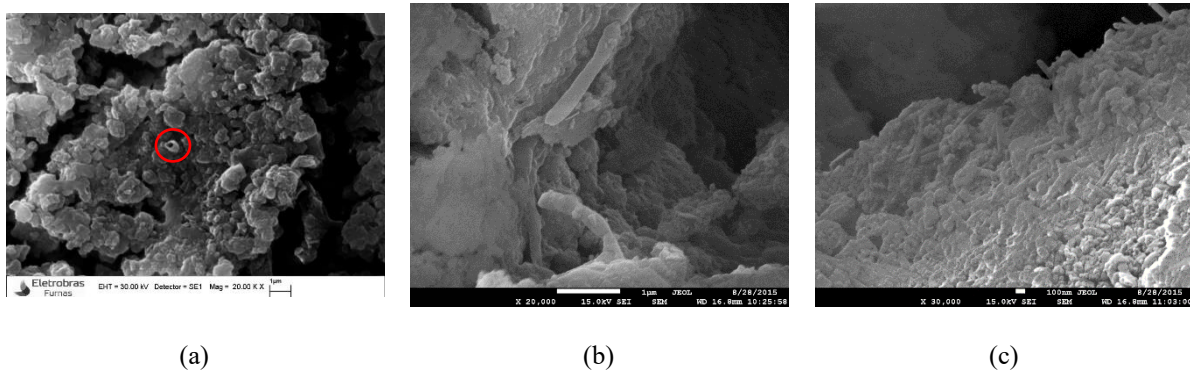
216 For porosimetry by adsorption and desorption of nitrogen, cubic specimens of approximately 7mm
217 were moulded from intact and compacted (A1 to A3, B1 to B3) samples, placed in the oven for 7 days
218 at a temperature of 105 °C, followed by freezing in liquid nitrogen (-195 °C) and drying in a vacuum
219 oven. The Nova 2200e porosimeter from Quantachrome instrument was used for this test.

220 For the MIP tests cubic specimens of approximately 1 cm³, also moulded from intact and compacted
221 (A1, A2, A3, B1, B2 and B3) samples, underwent freezing in liquid nitrogen (-195 °C) and drying in a
222 vacuum oven. MIP tests were performed using the AutoPore IV 9500 equipment from Micromeritics
223 Instrument Corporation.

224 **3. Mineralogical investigation**

225 The morphology and texture of the halloysite mineral vary according to the intensity of the
226 decomposition process and the effects of transport (Heiken and Wohletz, 1992), with the shape of
227 tubes being its predominant morphology (Senoussi et al., 2016). Allophanes, on the other hand, are
228 characterised by small hollow spheres of around 5 nm in diameter, 3.8 nm in internal diameter and
229 holes of around 0.3 nm (Iyoda et al., 2012).

230 SEM images of intact, A1 and B3 specimens are shown in **Fig. 3**. All three SEM images clearly indicate
 231 the existence of tubular structures that are characteristics of the halloysite mineral, while hollow
 232 sphere structures, characteristic of allophanes, could not be observed. In all three images, the
 233 diameter of the observed halloysite tubes range between 0.1 and 0.3 μm . A structural disorder and
 234 clay aggregations can also be observed in the three images.

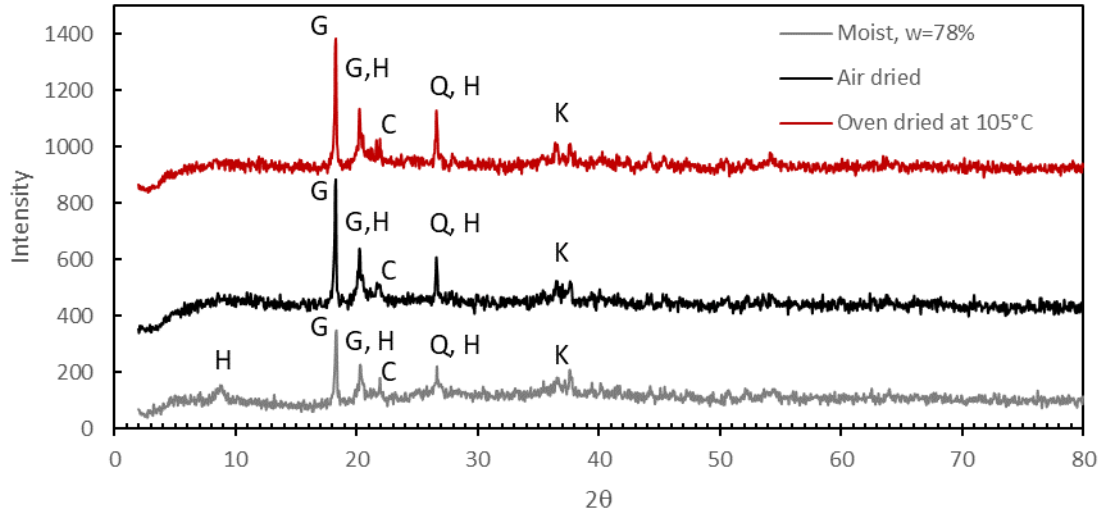


235 **Fig. 3.** SEM images of (a) Intact sample 20,000 \times magnification, where circle indicates a tubular structure, (b)
 236 A1, 20,000 \times magnification, and (c) B3, 30,000 \times magnification.

237 Failure to observe allophane in SEM images does not necessarily indicate that allophanes are not
 238 present in this material. Further tests need to be carried out to establish the presence or inexistence
 239 of allophane. Thus, **Fig. 4** shows the X-Ray Diffraction results of three specimens: oven dried at 105 $^{\circ}\text{C}$,
 240 air dried to the hygroscopic water content and at the natural water content of 78% without drying.
 241 The peaks identified in all three samples correspond to: gibbsite ($d = 4.84 \text{ \AA}$, $\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$), halloysite
 242 ($d = 4.36 \text{ \AA}$, $\text{Al}_2\text{Si}_2\text{O}_5(\text{OH})_4$), cristobalite ($d = 4 \text{ \AA}$, SiO_2), quartz ($d = 3.34 \text{ \AA}$, SiO_2), and kaolinite ($d = 2.4$
 243 \AA , $\text{Al}_2\text{Si}_2\text{O}_5(\text{OH})_4$) (Bordeepong et al., 2011; Deng et al., 2017; Falc3n et al., 2015; Joussein et al., 2006;
 244 Lizcano et al., 2006; Tchakout3e et al., 2020; Wesley, 1973). The main kaolinite peak ($d = 7.2 \text{ \AA}$) does
 245 not appear in any of the results, which indicates that kaolinite is poorly crystallized with disorder on
 246 the b axis (Hildebrando et al., 2009).

247 Halloysite is a hydrated polymorph of kaolinite (Emmanuel et al., 2019; Joussein et al., 2006). Based
 248 on its hydration state, halloysite is classified into two groups: the hydrated form (10 \AA), and the
 249 dehydrated form (7 \AA). The peak that characterises hydrated halloysite ($d = 10 \text{ \AA}$) (Cravero et al., 2016;
 250 Deng et al., 2017; Emmanuel et al., 2019; Wesley, 1973) can only be detected in the moist specimen.
 251 This finding is not surprising, since halloysite can dehydrate irreversibly at relative low temperatures
 252 (Bordeepong et al., 2011). However, it is widely reported in the literature (Bordeepong et al., 2011;
 253 Joussein et al., 2005) that this 10 \AA halloysite (hydrated) is likely to convert irreversibly into its 7 \AA
 254 (dehydrated) form. Nevertheless, the peak that characterises dehydrated halloysite ($d = 7 \text{ \AA}$,
 255 $\text{Al}_2\text{Si}_2\text{O}_5(\text{OH})_4$) (Joussein et al., 2005) is absent in all three specimens tested.

256 Since allophane is characterised by three main peaks at $d = 3.3 \text{ \AA}$, 2.25 \AA and 1.4 \AA , that cannot be
 257 observed in neither specimen, **Fig. 4** further confirms that allophane is not present in the material
 258 tested.

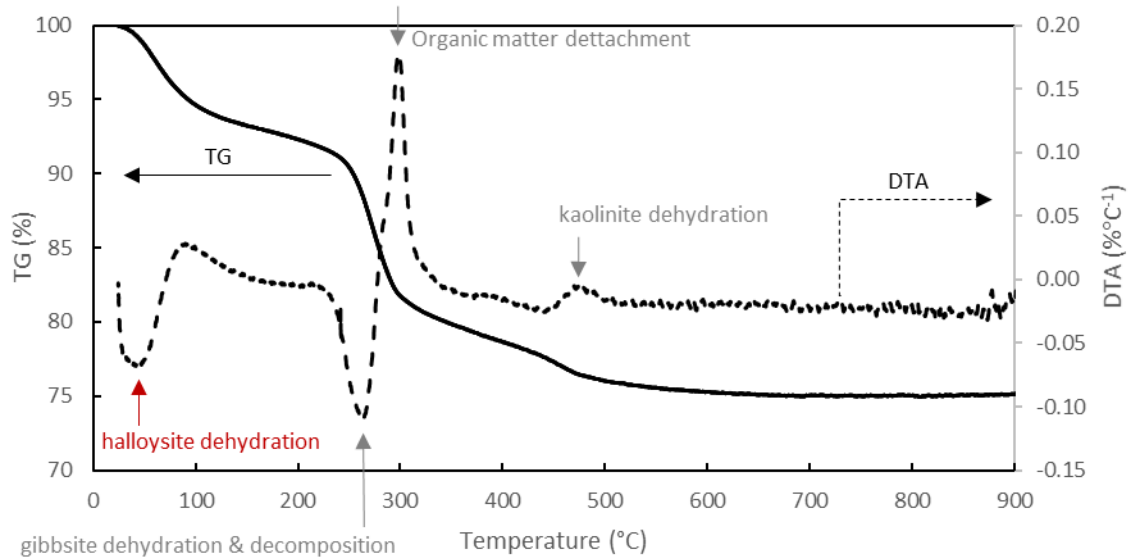


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260 **Fig. 4.** XRD of oven dried; air dried to hygroscopic water content and moist specimens, where H: halloysite, K:
 261 kaolinite, G: gibbsite, Q: quartz e C: cristobalite

262 To substantiate the XRD analyses, the weight-loss curve (TG) and its differential (DTA) for an intact
 263 specimen can be inspected in **Fig. 5**. The DTA of the specimen is characterised by four main thermal
 264 events: (1) an endothermic peak at around $50 \text{ }^\circ\text{C}$, corresponding to the loss of adsorbed water (surface
 265 and interlayer) (Iyoda et al., 2012; Singer et al., 2004) which characterises the dehydration of the
 266 halloysite according to Falcón et al. (2015); (2) a second endothermic peak at around $250 \text{ }^\circ\text{C}$, which
 267 corresponds to the dehydration and decomposition of the gibbsite (Falcón et al., 2015); (3) an
 268 exothermic peak at $300 \text{ }^\circ\text{C}$ caused by the detachment of the organic matter (Falcón et al., 2015); (4)
 269 and another exothermic peak at $470 \text{ }^\circ\text{C}$ which is the result of the dehydration of kaolinite (Falcón et
 270 al., 2015). From the DTA curve in **Fig. 5**, it can be observed that not much energy is required to remove
 271 the water from the halloysite, that is, it can occur at low temperatures.

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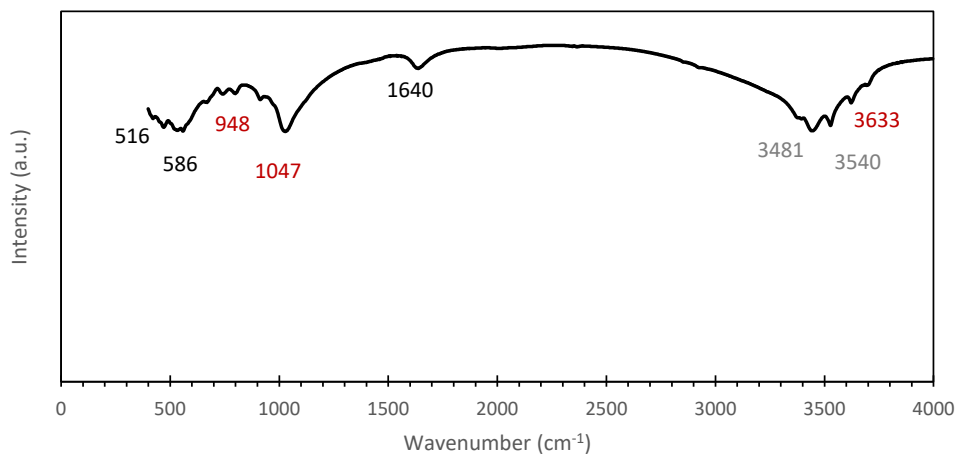


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274 **Fig. 5.** Thermal analyses curves (TG/DTA).

275 Complementarily to the XRD and TG/DTA analyses, FTIR test was performed on a specimen B (dried
 276 to the hygroscopic water content - **Fig. 6**). The absorption bands at 3633, 1047, 948 cm^{-1} correspond
 277 to halloysite (Espinoza and Meléndez, 2012; Singer et al., 2004); absorption bands at 3481, 3540 cm^{-1}
 278 correspond to gibbsite; strong OH stretching band of adsorbed water between approximately 3450–
 279 3550 cm^{-1} (Lun et al., 2014); absorption band at 1640 cm^{-1} relates to the hydroxyl group of water
 280 molecules present in the clay (H-O-H bending vibrations) (Polanský et al., 2017) specifically to the
 281 deformation of Si-OH bonds, which would be related to the presence of kaolinite (Espinoza and
 282 Meléndez, 2012).

283 The FTIR result confirms the presence of gibbsite, kaolinite and halloysite minerals identified in the
 284 XRD and thermal analyses. The wide spectrum of absorption bands of the gibbsite shows that the
 285 weathering process allowed the crystallization of this mineral (Espinoza and Meléndez, 2012).

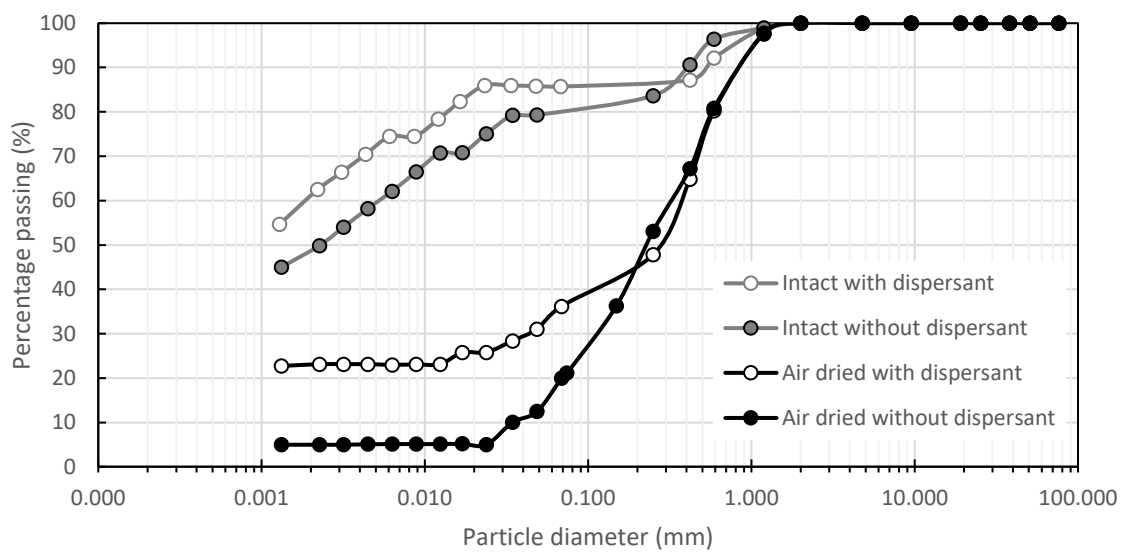


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287 **Fig. 6.** FTIR Spectra.

288 4. Geotechnical investigation

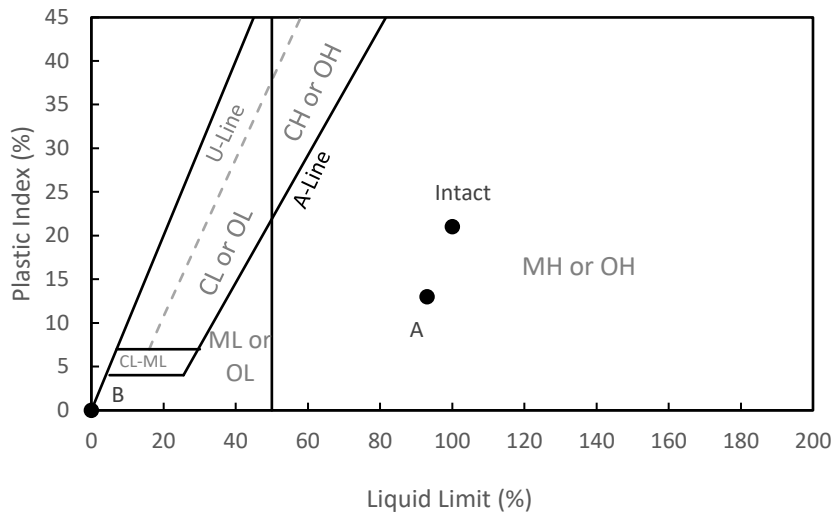
289 **Fig. 7** shows the particle size distribution of intact specimens and specimens prepared according to
 290 procedure B (air-dried to hygroscopic water content) with and without dispersant. The process of air
 291 drying to the hygroscopic water content forms aggregations of the finer particles (clay-sized)
 292 increasing the amount of sand-sized particles, a result similar to that obtained by Wesley (1973). When
 293 comparing the effect of dispersant on specimens prepared under the same conditions, it is observed
 294 that the dispersant separates particles, which generates an increase in the amount of silt- and clay-
 295 sized particles. However, the use of dispersant does not appear to be enough to erase the aggregation
 296 caused by the drying procedure.



297

298 **Fig. 7.** Particle size distribution of intact and air-dried samples with and without dispersant.

299 The plasticity of the material is influenced by the drying process, as demonstrated in the plasticity
 300 chart of **Fig. 8**. Specimens prepared according to the drying procedure B are non-plastic, while intact
 301 and specimens prepared according to drying procedure A are plastic. Although specimens A and intact
 302 are in the same region of the plasticity chart, there is a slight variation in the liquid limit and plasticity
 303 index obtained for these two types of specimens, with the intact specimen exhibiting higher liquid
 304 limit and plasticity index.



305

306 **Fig. 8.** Plasticity index chart

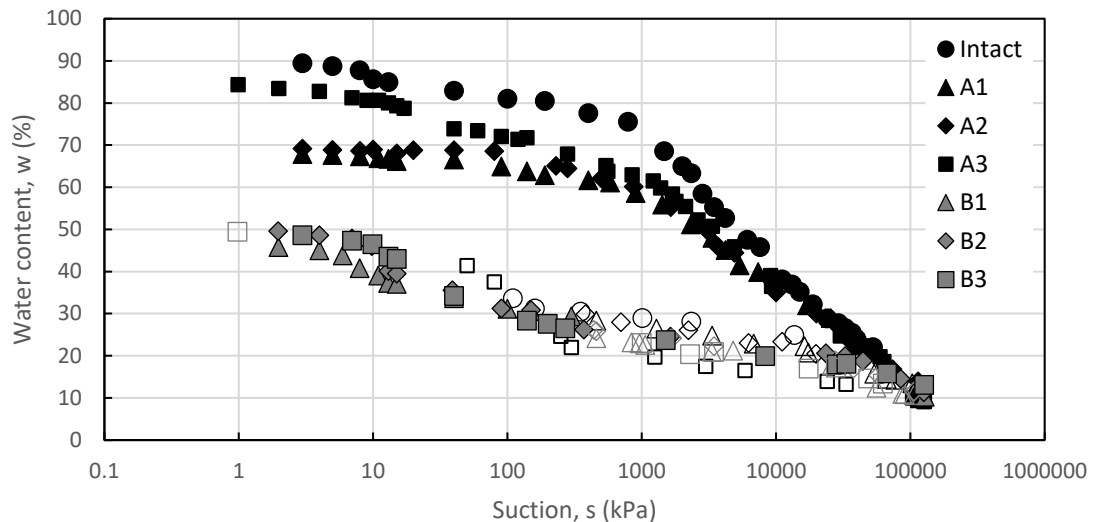
307 From the results of the particle size distribution and Atterberg limits, according to the Unified Soil
 308 Classification System (USCS), the material in an intact state and when it undergoes procedure A of
 309 drying can be classified as silt of high plasticity (MH-OH) and when it undergoes procedure B of drying,
 310 it is classified as silty sand (SM).

311 Thus, it is clear that the drying procedures influence the structural arrangement and consequently the
 312 classification according to particle size and the Atterberg limits. Similar results were obtained by
 313 Ferrari et al. (2013); Herrera (2005); Wesley (2009).

314 5. Hydromechanical investigation

315 5.1 Soil Water Retention Curve

316 Fig. 9 shows the results of the SWRC obtained from the drying and wetting paths on intact and
 317 compacted specimens. The drying preparation process has evident effects on three characteristics of
 318 the SWRC drying paths, that are: i) shape of the curves; ii) change in the air entry value; and iii) amount
 319 of water stored. While specimens prepared under drying procedure A, air-dried to the target water
 320 content values for saturated condition above 65%, specimens prepared under drying procedure B, air-dried to hygroscopic water content before
 321 wetting and compaction (B1, B2 and B3), present values close to 50%. In addition, for specimens
 322 prepared under drying procedure A, the variation in compaction water content seems to have greater
 323 influence than in specimens that have undergone drying procedure B, where no difference is observed
 324 between the three samples.
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327 **Fig. 9.** Soil Water Retention Curves, where closed symbols indicate drying paths and open symbols indicate
 328 wetting paths.

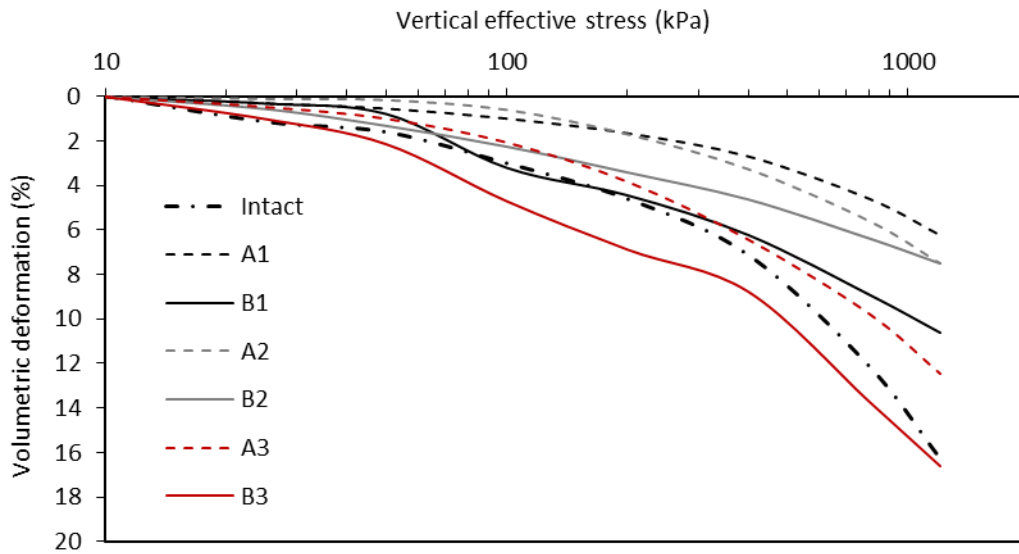
329 It is important to note that all A specimens (A1, A2 and A3) absorbed water after saturation (Fig. 9
 330 starting point of drying paths). However, specimen A3 absorbed considerably more water than the
 331 other two samples. This behaviour seems to contradict the literature on compacted soils, in the sense
 332 that soils compacted on the dry of optimum tend to expand more than soils compacted on the wet of
 333 optimum. This apparent contradiction may be a consequence of the fact that the three A specimens
 334 are already very close to saturation after compaction and probably have air bubbles occluded; these
 335 air bubbles are then more difficult to remove in specimens A1 and A2 than in specimen A3.”

336 Another interesting point highlighted by the SWRC is that specimens A present high hysteresis, while
 337 specimens B do not present any hysteresis between the drying and wetting paths. In the drying-
 338 wetting cycle, specimens A show that the drying process produces a change in how the water is stored.
 339 Additionally, it can be observed that the wetting curves of specimens A, B and intact are almost
 340 coincident.

341 When specimens reach suctions greater than 10 and 20 MPa, at water content close to 40% and 20%
 342 for the drying and wetting paths respectively, they begin to converge, which may imply that the air
 343 entry value of the mesopores (pores between 0.002 and 0.05 μm in diameter, according to IUPAC
 344 (1997) is the same regardless of the sample preparation process. Furthermore, despite having silt-
 345 sized particles the air entry values of intact and A specimens are high on the drying path.

346 **5.2. Consolidation test**

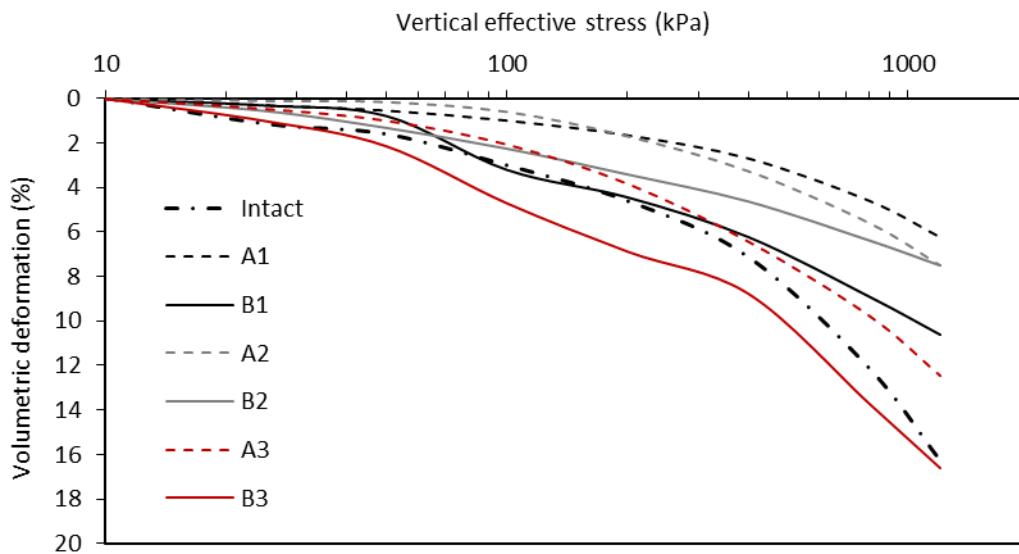
347 *The saturated compressibility curves are presented in*



348

349 **Fig. 10** in terms of volumetric deformation to remove the effect of the initial void ratio. Specimen B2
 350 (air dried to the hygroscopic water content) at optimum, is the least compressible specimen, while the
 351 intact specimen is the most compressible specimen (**Table 2**).

352 *Results presented in*



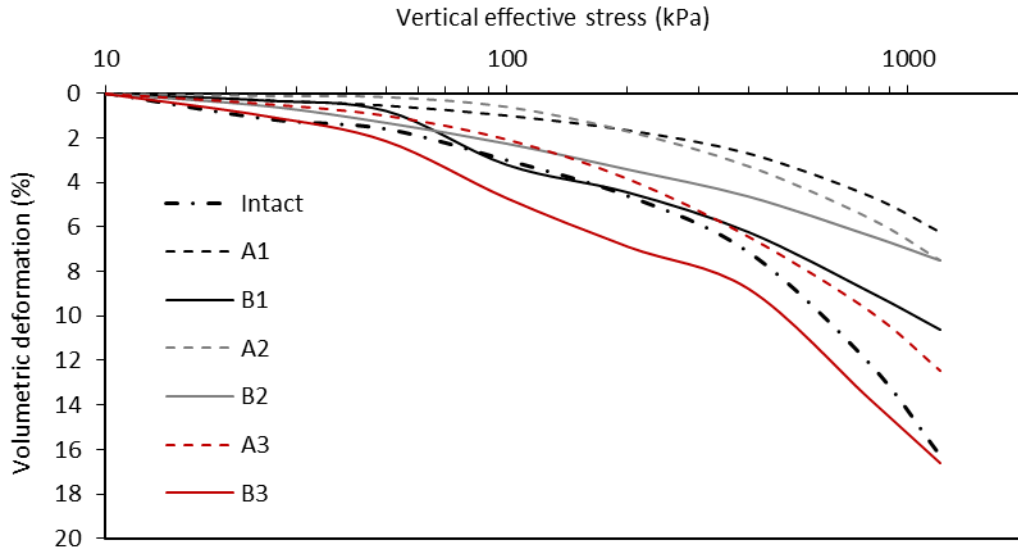
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354 **Fig. 10** do not provide conclusive evidence about the effects the drying procedures have on the
 355 compressibility of the material. However, compressibility tests indicate that compressibility increases
 356 in the following way: B2 < A1 < B1 < A2 < A3 < B3 < Intact (**Table 2**).

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358

359



360

361 **Fig. 10.** *Volumetric Deformation.*

362

363 **Table 2** *Compressibility parameters.*

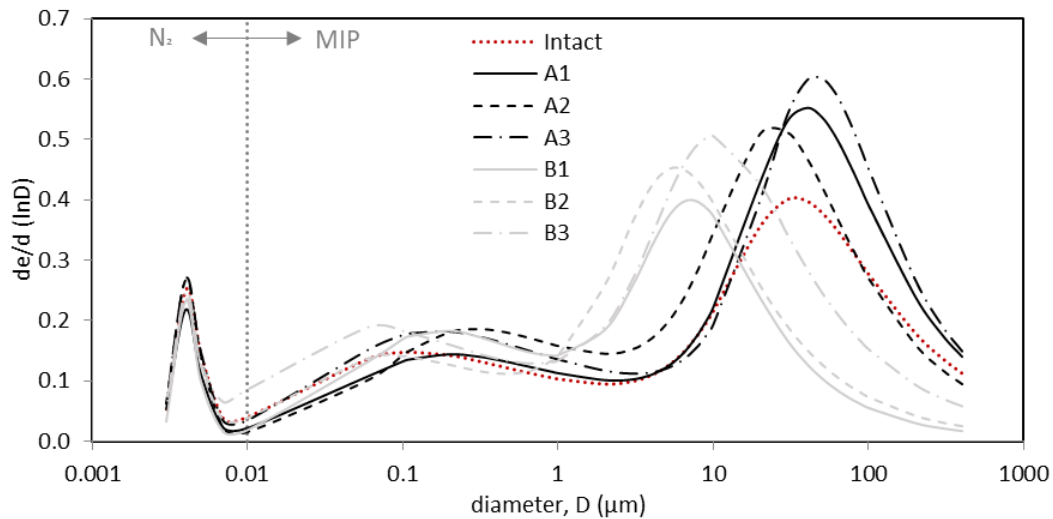
Sample	Water content, w (%)	Initial void ratio, e_i	Final void ratio, e_f	Compression index, C_c	Pre-consolidation stress, σ'_p (kPa)
Intact	78.0	2.95	2.43	0.754	250
A1	64.0	1.93	1.82	0.217	240
A2	65.3	1.78	1.64	0.246	190
A3	72.0	1.91	1.57	0.368	180
B1	38.0	1.42	1.24	0.223	92
B2	38.8	1.29	1.14	0.138	130
B3	42.0	1.42	1.03	0.398	180

364

365 **6. Structural investigation**

366 MIP and Nitrogen absorption tests were carried out in compacted and intact specimens. **Fig. 11** shows
 367 the Pore Size Distribution (PSD) curve obtained through the multi-modal fitting, similarly to that
 368 proposed by Lopes et al. (2014) of the combined experimental data. A trimodal distribution of pores
 369 can be observed in all specimens. According to IUPAC (1997) size classification, two of these peaks are
 370 observed in the macropores region, between 1 and 400 μm , and between 0.01 and 5 μm , and a third
 371 peak in the mesopores region, between 0.003 and 0.01 μm .

372 The drying and compaction processes do not seem to have affected the distribution of mesopores;
 373 while these processes appear to have played an important role in the distribution of macropores,
 374 particularly in the region of the largest pores.

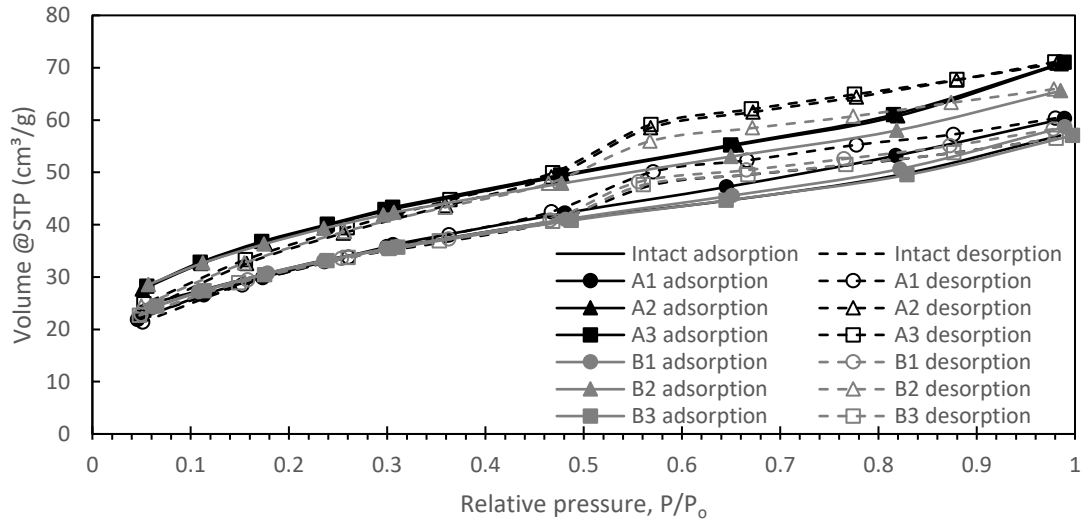


375

376 **Fig. 11.** Pore size distribution curves combining MIP and nitrogen adsorption (N_2) techniques.

377 In all cases, the void ratios obtained by the combined porosimetry tests were lower than the void
 378 ratios of the specimens presented in Table 1. This is a common issue reported by many researchers
 379 on porosimetry test carried out in different clayey soil samples (Cordão Neto et al., 2018; Kühn et al.,
 380 2021; Lopes, 2016; Pedrotti, 2016; Romero and Simms, 2008) and it could be mainly associated with
 381 (i) isolated pores surrounded by solids which are not intruded; (ii) some pores which are accessible
 382 only through smaller pores, i.e., they are not detected until smaller pores are penetrated, restricted
 383 porosity.

384 Isotherms were also obtained from Nitrogen adsorption and desorption tests. All isotherms belong to
 385 type IV with H3 hysteresis (**Fig. 12**) loops between adsorption and desorption curves (Gregg and Sing,
 386 1982; Sing and Williams, 2004; Yuan et al., 2012). According to Teixeira et al. (2001) when
 387 condensation occurs inside the pores, where the forces of attraction are greater due to the proximity
 388 between the molecules, it can occur at lower pressures than in non-porous solids. Evaporation,
 389 however, is hampered by the shape of the pore. Thus, the different paths characterise a hysteresis
 390 between the adsorption and desorption processes.



391

392 **Fig. 12.** Isotherms of adsorption and desorption of nitrogen.

393 Also from the Nitrogen adsorption test, the specific surface of the specimens could be calculated by
 394 the BET method (**Table 3**). The compacted specimens at optimum water content and the intact
 395 specimen exhibit the highest specific surface values.

396

Table 3 Specific surface of samples.

Sample	Specific surface (m ² /g)
Intact	126.04
A1	109.07
A2	130.22
A3	119.14
B1	108.06
B2	128.24
B3	107.58

397 **7. Discussions**

398 SEM images together mineralogical analyses of the volcanic ash under investigation did not reveal
 399 evidence that suggest allophane is present in the soil. However, these tests did indicate that halloysite
 400 is present in the soil composition. Halloysite tubes could be observed in SEM images of intact samples
 401 as well as samples subjected to drying procedure A, which involves air drying the disturbed soil up to
 402 the target water content required, and procedure B, which involves air drying the disturbed soil to the
 403 hygroscopic water content.

404 Based on its hydration state, halloysite is classified into two groups: the hydrated form ($d = 10 \text{ \AA}$), and
 405 the dehydrated form ($d = 7 \text{ \AA}$). Halloysite-(10 \AA) can easily and irreversibly dehydrate giving rise to
 406 halloysite-(7 \AA), because the interlayer water is weakly held. XRD test on moist samples (**Fig. 4**) shows

407 a peak at 10 Å and no peak at 7 Å, thus indicating the presence of hydrated halloysite. However, XRD
408 tests on air dried and oven dried (at 105 °C) samples did not reveal peaks at 10 Å or 7 Å. Nevertheless,
409 XRD test on all these three samples indicated other characteristic halloysite peaks ($d = 4.36$ Å and 3.35
410 Å). Furthermore, mineralogical tests also indicated the presence of halloysite in this soil.

411 The compaction behaviour changes completely depending on the drying procedure adopted (**Fig. 2**).
412 For performing standard compaction tests in the laboratory, samples collected are first subject to air
413 drying to the hygroscopic water content, and then the optimum water content and maximum dry unit
414 weight are determined. In contrast, when this material is compacted in the field, it is not first subjected
415 to drying to the hygroscopic water content. Instead, the natural water content is accessed and then
416 water is either added to the material, if this natural water content is lower than the optimum
417 determined in the lab, or the material is dried by revolving the soil, if the natural water content is
418 higher than the optimum. Popayán volcanic ash when compacted in laboratory following standard
419 drying to the hygroscopic water content presented the optimum water content and dry unit weight
420 of 38.8% and 11.31kN/m^3 , respectively, while when this material was compacted in the lab and dried
421 only to the target water content the optimum water content and dry unit weight obtained were 65.3%
422 and 9.32kN/m^3 , respectively. This demonstrates that compaction specifications determined in the lab
423 following standard procedures drive compaction conditions in the field excessively on the dry of
424 optimum. This in turn could inadvertently lead to undesired engineering properties.

425 This remarkable contrast in compaction behaviour has already been brought to light by previous
426 studies on volcanic soils (Hernandez et al., 2018; Wesley, 2009). These studies attributed the
427 difference in behaviour to the existence of either allophane and/or halloysite. According to the
428 authors the drying process irreversibly removes water at mineral scale, which then leads to a different
429 structure able to sustain less water.

430 Additionally, the loss of water at mineral level appears to have other consequences. The process of air
431 drying to the hygroscopic water content forms strongly held aggregations ($D_{50} = 0.002\text{mm}$ and
432 0.25mm for natural water content and air dried to hygroscopic water content respectively) that are
433 only marginally broken when dispersants are used (**Fig. 7**), while also changing the plasticity from
434 plastic (dried to target water contents) to non-plastic (dried to hygroscopic water content first) (**Fig.**
435 **8**).

436 Another interesting point highlighted by the SWRC (**Fig. 9**) is that intact and A (dried to target water
437 contents) specimens present high hysteresis, while B specimens (dried to hygroscopic water content)
438 do not present any hysteresis between the drying and wetting paths. This suggests that when water
439 leaves the mineral structure, it cannot be reincorporated in a wetting process. Specimens B, on the

440 other hand, had this loss of interlayer water anticipated during the sample preparation process.
441 Additionally, the fact that the wetting curves of specimens A, B and intact are almost coincident
442 reinforces the idea that the difference between the wetting and drying paths is mainly due to the
443 water leaving the mineral structure and not only the water leaving the pores (Inoue et al., 2012).
444 Therefore, it is reasonable to assume that the effect of the observed hysteresis is related to the loss
445 of the interlayer water (Kamble et al., 2012). This is another evidence, along with the change in
446 Atterberg Limits, which demonstrate the influence of interlayer water on soil behaviour.

447 Furthermore, despite having silt-sized particles the air entry values of intact and A specimens are high
448 on the drying path (Fig. 9). This behaviour is the opposite of what is expected, since intact and A
449 specimens have macropore diameters larger than specimens B (Fig. 11). Nevertheless, water remains
450 stored in intact and A specimen's pores for higher suction values than in B specimens. The presence
451 of hydrated halloysite may explain the great difference in the hydraulic properties of the soil,
452 influencing the water retention capacity (the water leaving the inside of the tubular structure of the
453 halloysite structure is an irreversible process) and affecting the strong connections between particles
454 (Nanzyo et al., 1989; Tadashi and SHoji, 2002; Wesley, 2009, 1973, 2014). Understanding whether this
455 difference relates to the water stored in pores or to the water stored inside the mineral is fundamental
456 to understand the hydromechanical behaviour of this material.

457 At first sight, results of the distribution of pores presented in Fig. 11 disagree with the particle size
458 distribution and SWRC (drying paths) results. Fig. 11 indicated that the drying procedure B, to the
459 hygroscopic water content, reduces the largest dominant pore size of the macropores (from 25 μm to
460 5 μm , average dominant macro pore size of group A and B respectively). Despite that, particle size
461 distribution results indicated that particles of samples that followed drying procedure B are larger in
462 diameter than the intact samples' particles, hence it can be automatically assumed that the pores of
463 B samples are larger than intact samples. Similarly, the suction at air entry values of B samples is
464 smaller than intact and A samples, which suggests once more that B samples have larger pores.

465 A more in-depth examination might help reconcile these results. It has been already established that
466 when halloysite is in its hydrated form it stores water in its interlayer. Once water drains out of the
467 interlayer it cannot be restored, hence the observations made via the SWRC, i.e.: (A) the air entry
468 value suction of intact and A samples is greater than in B samples in the drying paths; (B) the hysteresis
469 between the drying and wetting paths of intact and A samples; (C) the lack of hysteresis between the
470 drying and wetting paths of B samples; and (D) the fact that the curves of intact, A and B samples in
471 the wetting paths are almost coincident. A by-product of the water leaving permanently the halloysite
472 layers has been demonstrated by Kohyama et al. (1978). The authors showed evidence that the rolled
473 layers of tubular halloysite, originally tightly connected to each other, became separated as a result of

474 dehydration. Then, since halloysites (10 Å or 7 Å) are found in the form of aggregates from an early
475 stage in their formation (Jeong, 1998), the aggregates that contain halloysite particles increase in size,
476 hence the largest particle size of B samples observed in the particle size distribution. Following this
477 increase in aggregates size comes a reduction in the size of macropores, as observed in **Fig. 11**.

478 The drying procedures appear to have affected less the compressibility of the samples. The differences
479 observed are inconclusive. However, the high compressibility of the intact sample may be explained
480 by the loss of the original cemented structure. According to Lizcano et al. (2006) cementation of
481 volcanic soils is mainly given by the presence of minerals from the clay fraction, such as halloysite or
482 allophane. Wesley (2003) suggested that when these soils dry, the cemented structure deteriorates.

483 Similarly, the isolated analysis of the specific surface data presented in **Table 3** does not indicate any
484 special behaviour, since the range of variation could be considered a dispersion around the mean
485 (mean = 118.34m²/g and standard deviation = 10.05 m²/g). However, liquid limit and plasticity index
486 results (**Fig. 8**) demonstrate a wide variation in the way the particles interact with water, and this
487 behaviour is strongly influenced by the specific surface. This behaviour could be associated with the
488 loss of water (irreversible process) at mineral level.

489 8. Conclusions

490 Results of an experimental investigation carried out on a volcanic soil from Popayan, Colombia, were
491 presented here to investigate the extent to which changes in drying affect the engineering behaviour
492 of the soil. Mineralogical, geotechnical, hydromechanical, and structural tests were performed.

493 Mineralogical tests indicated the presence of halloysite, gibbsite, kaolinite, quartz and cristobalite in
494 the soil. From the geotechnical, hydromechanical and structural tests it was possible to recognise that
495 the drying procedure influenced: compaction behaviour; consistency limits; particle size distribution;
496 retention behaviour; compressibility and pore size distribution (> 1 µm), while not having great effect
497 on: specific surface; pore size distribution (< 1 µm).

498 Combined analyses of the laboratory tests performed suggested that the dehydration of halloysite at
499 mineral scale could then explain the distinct change in engineering behaviour observed when this soil
500 is subjected to drying.

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